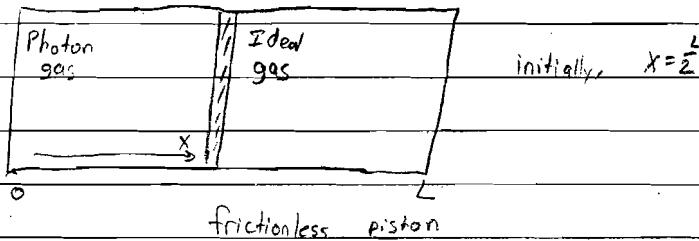


Jan 1997 #3 (SM)



a. both sides are in thermal equilibrium with a heat bath of temp T

$$\text{Photon gas: energy density } U = \frac{4}{c} \sigma T^4 \quad p = \frac{1}{3} U = \alpha U \quad (\alpha = \frac{1}{3})$$

$$(\sigma = 5.67 \cdot 10^{-8} \text{ J/m}^2 \cdot \text{K}^4)$$

$$\text{Ideal gas: } p = \frac{NkT}{V}$$

at constant temperature, $p = \frac{P_0(\frac{L}{2})}{L-x}$ $P_0 = \text{initial pressure} = 1 \text{ atm}$

$$\text{photon gas: initial pressure} = \frac{4}{3c} \sigma T^4 = 2 \cdot 10^{-6} \frac{\text{N}}{\text{m}^2} = 2 \cdot 10^{-11} \text{ atm}$$

at constant temperature, photon pressure is constant (independent of volume)

• at equilibrium with the heat bath, both gases stay at temp T

Equilibrium would occur when $P_{ph} = P_{gas}$, as x is varied, but this does not happen here. $P_{gas\text{-minimum}}$ is $\frac{P_0}{2} = 0.5 \text{ atm}$ while the photon gas pressure is $2 \cdot 10^{-11} \text{ atm}$.

• The piston moves all the way to $x=0$

b. No heat bath, insulating piston

$\Delta Q = 0$ for both gases (adiabatic)

$$dE_{ph} = -P_{ph} dV_{ph} \quad dE_{gas} = -P_{gas} dV_{gas} \quad dE_{ph} + dE_{gas} \neq 0$$

(In the quasistatic case, energy of the total system is not conserved, because by moving it quasistatically, the kinetic energy of the piston, which exists because $P_{ph} \neq P_{gas}$, so there is a net force, is continuously removed)

$$\text{Ideal gas, adiabatic: } P_{gas} V_{gas}^\gamma = \text{constant} \quad (\text{take } \gamma = \frac{5}{3})$$

$$V_{gas} = A \cdot (L-x) \Rightarrow P_{gas} (L-x)^\gamma = \text{const.} \Rightarrow P_{gas} \left(1 - \frac{x}{L}\right)^\gamma = \text{const.}$$

Photon gas adiabatic condition:

$$dE = -P dV$$

$$E = u(T)V \quad dE = \left(\frac{\partial E}{\partial T}\right)_V dT + \left(\frac{\partial E}{\partial V}\right)_T dV = \frac{\partial u}{\partial T} V dT + u dV$$

$$\frac{du}{dT} V dT + u dV = -P dV \quad P = \alpha u$$

$$\frac{d\ln u}{dT} = -(\alpha+1) \frac{u dV}{V} \quad \frac{du}{u} = -(\alpha+1) \frac{dV}{V}$$

$$\ln u = -(\alpha+1) \ln V + C$$

$$uV^{\alpha+1} = \text{const.} \quad P = \alpha u$$

$$P_{ph} V_{ph}^{\alpha+1} = \text{const} \quad V_{ph} = Ax$$

$$\Rightarrow P_{ph} \cdot \left(\frac{x}{L}\right)^{\alpha+1} = \text{const.}$$

$$\text{ideal gas: } P_{gas} (1-\frac{x}{L})^\gamma = P_0 \cdot \frac{1}{2^\gamma}$$

$$\text{photon gas: } P_{ph} \left(\frac{x}{L}\right)^{\alpha+1} = P_{ph=0} \cdot \frac{1}{2^{\alpha+1}}$$

$$\text{Let } u = \frac{x}{L}$$

$$P_{gas} = \frac{P_0}{2^\gamma (1-u)^\gamma} \quad P_{ph} = \frac{P_{ph=0}}{2^{\alpha+1} \cdot u^{\alpha+1}}$$

Equilibrium occurs when $P_{gas} = P_{ph}$

$$\frac{P_0}{2^\gamma (1-u)^\gamma} = \frac{P_{ph=0}}{2^{\alpha+1} u^{\alpha+1}} \quad P_0 u^{\alpha+1} = 2^{\gamma-(\alpha+1)} P_{ph=0} (1-u)^\gamma$$

$$P_0^{\frac{1}{\gamma}} u^{\frac{\alpha+1}{\gamma}} = 2^{\frac{1-\alpha-1}{\gamma}} P_{ph=0}^{\frac{1}{\gamma}} (1-u) \quad \left(\frac{P_0}{P_{ph=0}}\right)^{\frac{1}{\gamma}} \cdot 2^{\frac{\alpha+1}{\gamma}-1} u^{\frac{\alpha+1}{\gamma}} + u = 1$$

$\frac{P_0}{P_{ph=0}} \gg 1$, so the first term is much, much greater than the second term

$$u \approx \left(\frac{P_{ph=0}}{P_0}\right)^{\frac{1}{\gamma}} \cdot 2^{\frac{\alpha+1}{\gamma}-1} \quad \alpha = \frac{1}{3}, \gamma = \frac{5}{3} \quad \frac{5}{3}-1 = \frac{2}{3} = \frac{5}{4}-1 = \frac{1}{4}$$

$$u \approx \left(\frac{P_{ph=0}}{P_0}\right)^{\frac{3}{5}} 2^{-\frac{1}{4}}$$

$$u^* = \frac{x}{L} = 1.1 \cdot 10^{-8}$$

$$\text{photon gas temperature: } P_{ph} = \frac{P_{ph=0}}{2^{\alpha+1} u^{\alpha+1}} = .315 \text{ atm} = 3.15 \cdot 10^4 \text{ N/m}^2$$

$$P_{ph} = \frac{1}{3} u = \frac{4}{3} \frac{c}{c} T^4 = .315 \text{ atm} \quad T = \left(\frac{3c}{4c} \cdot .315 \text{ atm}\right)^{1/4}$$

$$T = 1.05 \cdot 10^5 \text{ K} \approx 9 \text{ eV}$$